

Thermodynamic Laws

Zeroth Law: When two systems are in thermal equilibrium with a third system, they must be in thermal equilibrium with each other.

First Law (closed system): $Q = m\Delta u + W / J$

Heat entering a system can either increase temperature (internal energy) or be used to perform work on the surroundings. It is the law of energy conservation, i.e., energy cannot be created or destroyed.

Second Law (isolated system): $m\Delta S_{total} \geq 0$

The entropy change of any system and its surroundings, considered together, is positive, and approaches zero for any process that approaches reversibility. It is considered the fundamental law of natural sciences.

The two classical statements of the Second Law:

Clausius statement: It is impossible to construct a device that operates in a cycle and produces no effect other than the transfer of heat from a cooler body to a hotter body.

Kelvin-Planck statement: It is impossible to construct a device that operates in a cycle and produces no effect other than the raising of a weight and the exchange of heat with a single reservoir.

Third Law: It is impossible to cool a body down to absolute zero.

Thermodynamic Laws (simplified)

First Law: You can't win, you can only break even.

Second Law: You can only break even at absolute zero.

Third Law: You can never reach absolute zero.

Thermodynamic Equations

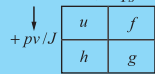
Ideal Gas Law: $pV = nRT = NkT$; $V = mv$; $n = m/M$; $k = R/N_A$

Helmholtz Function: $f = u - Ts$

The four thermodynamic potentials:

Gibbs Function: $g = h - Ts$

Equation of State: $p = -J \left(\frac{\partial f}{\partial v} \right)_T$



Enthalpy:

$$h = u + pv / J = \int \left[c_v + \frac{v}{J} \left(\frac{\partial p}{\partial T} \right) \right] dT + \frac{1}{J} \int \left[v \left(\frac{\partial p}{\partial v} \right)_T + T \left(\frac{\partial p}{\partial T} \right)_v \right] dv$$

$$\text{Entropy: } s = - \left(\frac{\partial f}{\partial T} \right)_v = - \left(\frac{\partial g}{\partial T} \right)_p = \int \frac{c_v dT}{T} + \frac{1}{J} \int \left(\frac{\partial p}{\partial T} \right)_v dv$$

$$\text{Isochoric Specific Heat: } c_v = \left(\frac{\partial u}{\partial T} \right)_v = T \left(\frac{\partial s}{\partial T} \right)_v = c_v^0 + \int_{T^0}^T \left(\frac{\partial^2 p}{\partial T^2} \right)_v dT$$

$$\text{Isobaric Specific Heat: } c_p = \left(\frac{\partial h}{\partial T} \right)_p = T \left(\frac{\partial s}{\partial T} \right)_p = c_v - \frac{T \left(\frac{\partial p}{\partial v} \right)_T^2}{J \left(\frac{\partial^2 p}{\partial v^2} \right)_T}$$

Specific Heat Ratio: $\gamma = c_p / c_v = \kappa / \kappa_s$

Velocity of Sound:

$$a = v \sqrt{g_c \gamma \left(\frac{\partial^2 f}{\partial v^2} \right)_T} = v \sqrt{-g_c \gamma \left(\frac{\partial p}{\partial v} \right)_T} = v \sqrt{g_c \left[\frac{T \left(\frac{\partial p}{\partial T} \right)_v^2}{J c_p} - \left(\frac{\partial p}{\partial v} \right)_T \right]}$$

Nomenclature

a	velocity of sound	ft/sec	m/sec
c_p	isobaric specific heat	Btu/lbm ^o -R	kJ/kg ^o -K
c_v	isochoric specific heat	Btu/lbm ^o -R	kJ/kg ^o -K
f	Helmholtz function	Btu/lbm	kJ/kg
g	Gibbs function	Btu/lbm	kJ/kg
g_c	gravitational conversion factor	32.174 lbm-ft/lb _m -sec ²	1.0
h	enthalpy	Btu/lbm	kJ/kg
J	Joule's constant	778.16926 ft-lb _m /Btu	1.0
m	mass	lb _m	kg
M	molecular weight	---	---
N	no. of molecules	---	---
n	no. of moles	lb _m /mol	kg/mol
p	pressure	lb _f /ft ²	kPa
Q	heat	Btu	kJ
s	entropy	Btu/lbm ^o -R	kJ/kg ^o -K
T	temperature	°R	°K
u	internal energy	Btu/lbm	kJ/kg
v	specific volume	ft ³ /lb _m	m ³ /kg
V	volume	ft ³	m ³
w	work	ft-lb _f	kJ
κ	isothermal compressibility	ft ³ /lb _f	(kPa) ⁻¹
κ_s	adiabatic compressibility	ft ³ /lb _f	(kPa) ⁻¹
γ	specific heat ratio	---	---

Superscript: 0 = heat capacity at zero pressure

(Physical Constants on reverse side.)

Carrying Capacity of Refrigeration Lines - Tons of Refrigeration

R-22	Line Size		Suction Lines				Liquid Lines				Discharge Lines							
	Type L Copper OD (Inches)	Velocity 1500 fpm	ΔT = 1°F		ΔT = 0.39 psi		ΔT = 0.57 psi		ΔT = 0.81 psi		ΔT = 1.1 psi		Velocity 150 fpm		ΔT = 1°F, Δp = 1.9 psi		ΔT = 1°F, Δp = 3.0 psi	
			Suction Temperature (°F)	-40	-20	0	20	Suction Temperature (°F)	-40	-20	0	20	Suction Temperature (°F)	-20	20	20	20	Suction Temperature (°F)
3/8	0.09	0.14	0.22	0.33	0.02	0.04	0.06	0.09	0.09	0.27	2.40	0.94	0.99	0.38	0.42	0.80	0.87	1.00
1/2	0.17	0.27	0.43	0.64	0.06	0.11	0.18	0.28	0.35	0.98	3.58	1.28	1.33	0.72	0.78	1.25	1.35	1.50
5/8	0.26	0.41	0.64	0.98	0.11	0.18	0.28	0.43	0.55	1.54	5.23	1.83	1.88	1.10	1.17	1.95	2.08	2.30
7/8	0.36	0.56	0.91	1.42	0.13	0.23	0.30	0.49	0.76	2.13	7.19	2.43	2.48	1.15	1.19	2.45	2.60	2.90
1-1/8	0.46	0.76	1.26	2.02	0.16	0.26	0.36	0.56	0.89	2.90	9.54	3.13	3.18	1.25	1.29	2.75	2.90	3.25
1-3/8	0.56	0.96	1.56	2.42	0.18	0.28	0.38	0.58	0.94	3.70	12.10	3.37	3.42	1.30	1.34	2.85	3.00	3.35
1-5/8	0.66	1.06	1.66	2.62	0.20	0.30	0.40	0.60	0.97	4.50	14.60	3.54	3.59	1.35	1.39	2.95	3.10	3.45
2-1/8	0.76	1.26	1.86	2.92	0.23	0.33	0.43	0.63	1.00	5.50	17.60	3.73	3.78	1.40	1.44	3.05	3.20	3.55
2-3/8	0.86	1.36	1.96	3.18	0.25	0.35	0.45	0.65	1.05	6.50	20.60	3.92	3.97	1.45	1.49	3.15	3.30	3.65
3-1/8	0.96	1.46	2.06	3.44	0.27	0.37	0.47	0.67	1.07	7.50	23.60	4.11	4.16	1.50	1.54	3.25	3.40	3.75
3-3/8	1.06	1.56	2.16	3.70	0.29	0.39	0.49	0.69	1.09	8.50	26.60	4.30	4.35	1.55	1.59	3.35	3.50	3.85
4-1/8	1.16	1.66	2.26	3.94	0.31	0.41	0.51	0.71	1.11	9.50	29.60	4.49	4.54	1.60	1.64	3.45	3.60	3.95
4-3/8	1.26	1.76	2.36	4.18	0.33	0.43	0.53	0.73	1.13	10.50	32.60	4.68	4.73	1.65	1.69	3.55	3.70	4.05
5-1/8	1.36	1.86	2.46	4.42	0.35	0.45	0.55	0.75	1.15	11.50	35.60	4.87	4.92	1.70	1.74	3.65	3.80	4.15
6-1/8	1.46	1.96	2.56	4.66	0.37	0.47	0.57	0.77	1.17	12.50	38.60	5.06	5.11	1.75	1.79	3.75	3.90	4.25

Suction Line capacities based on 200 equivalent feet, saturated refrigerant vapor, 80°F liquid (R-717)
 Liquid Line capacities based on 200 equivalent feet, saturated refrigerant liquid (R-717)
 Discharge Line capacities based on 100 equivalent feet, 100°F condensing temperature, 25°F discharge vapor temperature, 25°F superheated vapor entering compressor, 80°F liquid (R-717)
 All line capacities calculated with thermodynamic and thermophysical properties computed using NIST Refprop v8.01 database
 Approximate line capacity for other capacities and equivalent lengths: Line capacity = Table capacity * (Table capacity / Actual capacity)
 Approximate saturation temperature ΔT for other line capacities and equivalent lengths: ΔT = Actual ΔT / Table ΔT * Actual capacity / Table capacity

Thermodynamic Terms

Types of Energy
potential: energy due to relative position.
kinetic: energy due to relative velocity.
internal: the sum of all potential and kinetic energies.
heat: energy transferred between system and surroundings due to temperature difference.
work: energy transferred between system and surroundings due to pressure difference.

Heat Units
calorie: energy required to raise one gram of water by 1°C.
BTU: energy required to raise one pound of water by 1°F.

Thermodynamic Systems
closed: energy can cross system boundaries, but mass cannot.
open: both energy and mass can cross system boundaries.
isolated: neither energy nor mass can cross system boundaries.

States
saturation pressure (vapor pressure): pressure at which a phase change will take place at a given temperature.
saturation temperature: temperature at which a phase change will take place at a given pressure.
subcooled liquid: liquid at a temperature below its saturation temperature.
superheated vapor: vapor at a temperature above its saturation temperature.
critical point: saturated state where the liquid and vapor phases become indistinguishable.
triple point: saturated state where the solid, liquid and vapor phases are in equilibrium.

Thermodynamic Properties
mass: the property of a body that is a measure of its inertia, causing it to have weight in a gravitational field.
temperature: a measure of the relative hotness or coldness of a substance.
pressure: force per unit area.
density: the amount of mass of a substance within a unit volume.
specific volume: the amount of space occupied by a unit mass of a substance.
internal energy: sum of all potential and kinetic energies in a substance.
enthalpy: the total useful energy in a substance.
entropy: a measure of energy which is no longer available to perform useful work.
specific heat: the amount of energy necessary to raise the temperature of a substance under constant pressure (Cp) or constant volume (Cv).
latent heat: energy necessary to produce a change in phase without causing a temperature change.
sensible heat: energy that changes temperature.

Thermodynamic Processes
adiabatic: no heat crosses the system boundary
isenthalpic (throttling): constant enthalpy
isotropic: constant entropy
isobaric: constant pressure
isochoric (isometric): constant volume
isothermal: constant temperature

Idealized Thermodynamic Cycles
Vapor Cycles
Carnot: Isentropic compression; isobaric/isothermal heat addition (vaporization); isentropic expansion; isobaric/isothermal heat rejection (condensation). This cycle is closely approximated in actual steam turbine plants.
Rankine: Isentropic compression; isobaric heat addition; isobaric/isothermal heat addition (vaporization); isentropic expansion; isobaric/isothermal heat rejection (condensation). This cycle is closely approximated in actual steam turbine plants.

Combustion Cycles
Otto: Isentropic compression; isochoric heat addition; isentropic expansion; isochoric heat rejection.
Diesel: Isentropic compression; isobaric heat addition; isentropic expansion; isochoric heat rejection.
Brayton: Isentropic compression; isobaric heat addition; isentropic expansion; isobaric heat rejection.
Ericsson: Isentropic compression; isobaric heat addition; isothermal expansion; isobaric heat rejection.
Stirling: Isothermal compression with heat rejection; isochoric heat addition; isothermal expansion with heat addition; isochoric heat rejection.

Refrigeration Cycles
Reversed Carnot: Isentropic compression; isobaric/isothermal heat rejection (condensation); isentropic expansion; isobaric/isothermal heat addition (evaporation).
Vapor Compression: Isentropic compression; isobaric heat rejection; isobaric/isothermal heat rejection (condensation); isenthalpic expansion; isobaric/isothermal heat addition (evaporation). This is the idealized cycle for most modern air conditioning and refrigeration systems.
Reversed Brayton: Isentropic compression; isobaric heat rejection; isentropic expansion; isobaric heat addition. This cycle uses air as a refrigerant.
Reversed Stirling: Isothermal compression with heat rejection; isochoric heat addition; isothermal expansion with heat addition; isochoric heat rejection.

Fluid Flow Equations

Bernoulli equation:

$$\frac{P_1 \rho_c}{\gamma} + \frac{V_1^2}{2g} + z_1 = \frac{P_2 \rho_c}{\gamma} + \frac{V_2^2}{2g} + z_2; \quad \gamma = \rho g$$

Reynolds Number:

$$R_e = \frac{vD}{\nu} = \frac{vD\rho}{\mu g_c}$$

Colebrook's equation:

$$\frac{1}{\sqrt{f}} = -2 \log_{10} \left[\frac{\epsilon/D}{3.7} + \frac{2.51}{R_e \sqrt{f}} \right]$$

Darcy-Weisbach equation:

$$h_f = f \frac{L_c}{D} \frac{v^2}{2g}$$

Orifice equation (incompressible flow):

$$m = C_f A_o \sqrt{2g \rho \Delta P}; \quad R_e > 250$$

Orifice equation (vapor flow):

$$m = C_f A_o \sqrt{\frac{2k}{k-1} P_o g_c v_o \left(\frac{P_d}{P_o} \right) \left[1 - \left(\frac{P_d}{P_o} \right)^{\frac{(k-1)}{k}} \right]}$$

where $\frac{P_d}{P_o} >$ critical pressure ratio

Critical pressure ratio:

$$\frac{P_d}{P_o} = \left(\frac{2}{k+1} \right)^{\frac{k}{k-1}}$$

Nomenclature

	I-P	S-I
A_o	orifice area	ft ²
C_f	flow coefficient	---
D	diameter	ft
f	friction factor	---
g_c	gravitational conversion factor	32.174 lbm-ft/lbf-sec ²
k	specific heat ratio	---
h_f	head loss due to friction	ft
L_c	equivalent length	ft
m	mass flow rate	lbm/sec
p	pressure	lbf/ft ²
R_e	Reynold's Number	---
v	velocity	ft/sec
z	height	ft
ϵ	effective roughness	ft
γ	specific weight	lbm/ft ³ -sec ²
μ	absolute viscosity	lbf-sec/ft ²
ν	kinematic viscosity	ft ² /sec
ρ	density	lbm/ft ³

Subscripts: m = upstream; d = downstream

Physical Constants

	I-P	S-I
g	standard acceleration due to gravity	32.1740 ft/sec ²
k	Boltzmann's constant	5.657306x10 ⁻²⁴ ft-lbf/R
N_A	Avogadro's constant	2.7151056x10 ²³ lb/mol
R	universal gas constant	1545.349 ft-lbf/lb-mol-R

Refrigerant Data

ASHRAE Number	Chemical Name	Sporlan Letter Designation	Color Designation (PMS No.)	ASHRAE 34 Safety Group	CAS Registry Number	Critical Values ² Temperature (°F)	Pressure (psia)	Specific Vol (ft ³ /lb)	Molecular Weight
R-10	tetrachloroethane (carbon tetrachloride)			B1	56-23-5	542.03	861.27	0.0287	153.82
R-11	trichlorofluoromethane	H	Orange (021)	A1	75-29-4	538.33	639.27	0.029	137.17
R-12	dichlorodifluoromethane	F	White (N/A)	A1	75-71-8	233.55	599.89	0.0284	120.91
R-12B1	bromochlorodifluoromethane (halon 1211)			A1	353-59-3	398.94	594.94	0.0225	165.36
R-13	trichlorofluoromethane	E	Light Blue (2975)	A1	75-79-9	83.71	592.31	0.0275	104.46
R-13B1	bromotrifluoromethane (halon 1301)	T	Pinkish-Red/Coral (177)	A1	75-63-8	152.80	574.90	0.0215	148.91
R-14	tetrafluoroethane (carbon tetrafluoride)			A1	75-73-9	-50.15	543.89	0.0256	88.01
R-20	trichloroethane (chloroform)			A1	75-43-4	504.23	734.61	0.0234	119.39
R-21	dichlorodifluoroethane	B1	75-43-4	B1	75-43-4	353.21	751.30	0.0306	102.92
R-22	chlorodifluoroethane	V	Light Green (352)	A1	75-45-6	205.06	723.74	0.0206	86.47
R-22B1	bromochlorodifluoroethane (halon 1201)			A1	281-88-8	214.33	744.32	0.0206	120.92
R-23	trifluoroethane	G	Light Blue-Gray (428)	A1	75-46-7	78.66	701.40	0.0305	70.01
R-30	dichloromethane (methylene chloride)	B2	75-46-7	B2	75-46-7	455.27	881.83	0.037	84.93
R-31	chlorodifluoroethane	B1	75-46-7	B1	593-70-4	335.17	744.20	0.0275	84.48
R-32	difluoroethane (methylene fluoride)	A2	75-10-5	A2	75-10-5	172.59	638.61	0.0378	52.02
R-40	chloromethane (methyl chloride)	B2	74-87-3	B2	74-87-3	289.49	665.95	0.0446	50.49
R-410	fluoromethane (methyl fluoride)			A1	593-53-3	111.43	655.29	0.0606	34.03
R-50	methane	A3	74-82-8	A3	-118.70	-118.70	666.40	0.098	16.04
R-60	hexachloroethane			B3	67-72-1	808.3	571	0.0284	236.74
R-113	1,1,2-trichloro-1,2,2-trifluoroethane			A1	76-13-1	417.31	492.00	0.0286	167.37
R-134	1,2-dichloro-1,1,2,2-tetrafluoroethane	B	Dark Blue/Navy (302)	A1	76-14-2	294.22	472.39	0.0276	170.92
R-115	chloropentafluoroethane	A1	175-51-3	A1	76-15-3	452.52	602.61	0.0261	154.47
R-116	hexafluoroethane			A1	76-16-4	67.78	441.20	0.0258	138.01
R-122	2,2-dichloro-1,1,1-trifluoroethane			B1	306-83-2	262.63	531.10	0.0291	152.93
R-124	2-chloro-1,1,1,2-tetrafluoroethane	M	Deep Green (335)	A1	2837-89-0	252.10	525.66	0.0286	136.48
R-125	pentafluoroethane			A1	334-33-6	151.12	526.34	0.0280	120.02
R-132B1	1,1,1,3,3,3-hexafluoroethane			A1	75-25-5	720.5	841.6	0.0283	147.16
R-134a	1,1,1,2-tetrafluoroethane	J	Light Blue (2975)	A1	811-97-2	213.91	588.75	0.0313	102.03
R-141b	1,1-dichloro-1-fluoroethane			A1	1717-00-6	399.56	616.41	0.0348	116.95
R-142b	1-chloro-1,1-difluoroethane			A2	75-46-3	278.76	597.89	0.0388	100.50
R-143a	1,1,1-trifluoroethane			A2	429-46-2	183.20	547.60	0.0370	84.04
R-150	1,2-dichloroethane			A2	107-06-2	55.1	778.9	0.0384	98.96
R-150a	1,1-dichloroethane			A2	75-34-3	481.7	735.3	0.038	96.95
R-152a	1,1-difluoroethane			A2	75-07-6	228.07	655.10	0.0435	69.69
R-160	chloroethane (ethyl chloride)			A3	75-00-3	369.1	764	0.0364	64.51
R-160B1	ethyl bromide (halon 2001)			A3	74-48-4	447.3	903.6	0.0356	108.96
R-161	fluoroethane (ethyl fluoride)			A3	333-85-5	216.9	662	0.036	48.86
R-170	ethane	K		A3	74-84-0	89.2	705.59	0.1120	30.07
R-227ea	1,1,1,2,3,3,3-heptafluoropropane			A1	431-89-0	217.04	622.21	0.0276	170.03
R-232ea	1,1,1,2,3,3,3-heptafluoropropane			A1	431-89-0	217.04	622.21	0.0276	170.03
R-232fa	1,1,1,3,3,3-hexafluoropropane			A1	690-39-1	256.86	464.12	0.0291	152.04
R-242ca	1,1,2,2,2-pentafluoropropane			A1	678-98-7	345.96	569.27	0.0306	134.05
R-242fa	1,1,1,3,3,3-pentafluoropropane			A1	409-73-11	209.28	527.94	0.0318	124.05
R-C270	cyclopentane			A3	75-19-4	257.27	809.23	0.0617	42.08
R-290	propane			A3	74-88-6	206.06	616.07	0.0706	44.10
R-C318	octafluorocyclobutane			A1	115-25-3	239.41	402.84	0.0278	200.04
R-600a	n-butane			A3	106-97-8	305.62	509.66	0.0375	58.12
R-600a	2-methyl propane (isobutane)			A3	75-28-5	274.46	527.94	0.0714	58.12
R-601	n-pentane			A3	109-66-0	305.8	467	0.0630	72.15
R-601a	2-methyl butane (isopentane)			A3	76-34-4	370.0	490	0.0679	72.15
R-610	ethyl ether			A3	60-29-7	380.57	527.94	0.0669	74.12
R-611	methyl formate			B2	107-31-3	416.29	670.23	0.0459	60.05
R-612	ethyl amine			B2	74-89-5	314.4	1062.0	0.0714	31.06
R-631	ethyl amine			A1	75-04-7	361.4	815.11	0.0832	45.08
R-702a	normal hydrogen			A3	1332-74-0	-399.93	190.75	0.5230	2.02
R-704	helium			A1	744-61-7	-452.31	4.00	0.0029	4.00
R-717	ammonia	A	Silver	A3	7668-91-7	269.89	1943.7	0.1643	17.03
R-718	water/steam			A1	7732-18-5	705.18	3199.2	0.0497	18.02
R-720	neon			A1	7448-01-9	-379.58	388.50	0.0332	21.18
R-728	nitrogen			A1	7727-37-9	-252.52	572.81	0.8710	28.01
R-729	air					-220.86	548.97	0.0529	28.97
R-732	oxygen			A1	7782-44-7	-181.42	731.42	0.0367	32.00
R-740	argon			A1	7449-39-1	-188.48	714.68	0.032	39.95
R-744	carbon dioxide			A1	124-38-9	87.76	1069.9	0.0342	44.01
R-744A	nitrous oxide			A1	10028-97-2	97.61	1050.1	0.0353	44.01
R-764	sulfur dioxide			B1	7448-09-5	315.77	1142.9	0.0305	64.06
R-1120	trichloroethane (trielene)			B1	78-01-6	568	712	0.0312	131.39
R-1130	trans-1,2-dichloroethane (dielene)			B3	156-60-5	469.9	795	0.035	96.95
R-1132a	1,2-difluoroethane (vinylidene fluoride)			A1	75-36-7	65.46	646.62	0.0265	64.04
R-1140	chloroethene (vinyl chloride)			B3	75-01-4	305.3	746.9	0.035	62.50
R-1150	ethene (ethylene)			A3	74-85-1	485.5	731.00	0.0748	28.05
R-1270	propene (propylene)			A3	115-07-1	198.36	676.54	0.0717	42.08

Refrigerant Blends

ASHRAE Number	Composition (mass percentages)	Sporlan Letter Designation	Color Designation (PMS No.)	ASHRAE 34 Safety Group	Trade Name	Critical Values ² Temperature (°F)	Pressure (psia)	Specific Vol (ft ³ /lb)	Molecular Weight
R-401A	R-22/R-152a/R-124 (53/13/34)	X	Pinkish-Red/Coral (177)	A1	MP39	221.48	689.03	0.0324	94.44
R-401B	R-22/R-152a/R-124 (61/17/28)	F	Yellow-Brown/Mustard (124)	A1	MP66	218.28	679.07	0.0322	92.84
R-402A	R-125/R-290/R-2 (80/2/18)	L	Light Brown/Gand (461)	A1	HP18	182.85	614.11	0.0241	101.55
R-402B	R-125/R-290/R-2 (38/2/60)	R	Green-Brown/Olive (385)	A1	HP81	181.45	656.35	0.0299	94.71
R-402A	R-22/R-290/R-218 (5/75/20)			A1	1SC20N [®] 69-S	196.2	680	0.032	91.99
R-404A	R-22/R-220/R-142b (65/6/28)			A1	1SC20N [®] 69-L	191.7	638	0.032	92.36
R-404A	R-125/R-134a/R-134a (44/52/4)	S	Orange (021)	A1	HPR2, FX-70	161.86	541.71	0.0328	97.60
R-405A	R-22/R-152a/R-142b/R-C318 (49/75/5/42.5)			A1	GREENCOOL G2015	222.8	622	0.032	111.91
R-405B	R-22/R-600a/R-142b (55/41/1)			A2	GHG	241.7	708	0.032	100.26
R-406B	R-22/R-600a/R-142b (65/4/31)			A2	GHG-HP			0.032	88.58
R-407A	R-32/R-125/R-134a (20/40/40)	V	Line Green (388)	A1	Klea [®] 69	179.44	650.77	0.0302	90.11
R-407B	R-32/R-125/R-134a (18/70/28)			A1	Klea [®] 61	185.06	692.21	0.0295	102.94
R-407C	R-32/R-125/R-134a (23/25/52)	N	Medium Brown (471)	A1	Klea [®] 66, Sava 9000	186.89	672.11	0.0312	86.20
R-407D	R-32/R-125/R-134a (15/15/70)			A1	Dark Brown/Chocolate (4				